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## Key indicators

Single-crystal X-ray study
$T=123 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.005 \AA$
$R$ factor $=0.045$
$w R$ factor $=0.119$
Data-to-parameter ratio $=7.7$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## $N$-(3-Nitrophenyl)maleimide

The title compound, $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$, belongs to a series of N -arylmaleimides, which can be used as photoinitiators for free-radical polymerization. The dihedral angles between the planes of the benzene and imide rings are 56.2 (1) and $52.9(1)^{\circ}$ in the two independent molecules in the asymmetric unit.

## Comment

There is considerable activity related to the use of $N$-substituted maleimides as photoionizers for free-radical polymerization, where the maleimide can produce the initiating radical species (Pyriadi \& Nabeel, 1988; Andersson et al., 1996; Hoyle et al., 1999). In continuing the structural studies on N -substituted maleimide systems, to study the behaviour of $\mathrm{C}_{\text {aryl }}-\mathrm{N}$ distance and imide/benzene interplanar angle, the crystal structure determination of $m$-nitrophenylmaleimide, $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$, (I), was undertaken. The reactivity of N -aromatic maleimides in photopolymerization processes as a function of the angle between the maleimide and benzene rings has been analysed (Miller et al., 2000). The p-nitrophenylmaleimide ( $p$-NPM) system has been reported by our research group (Moreno-Fuquen et al., 2003). This structure has a close analogy to the title compound and it has been used as a model for comparison.

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A perspective view of the two independent molecules in the asymmetric unit of the title compound, showing the atomic numbering scheme, is given in Fig. 1. Focusing on the $\mathrm{N}-\mathrm{C}_{\text {aryl }}$ bond length, in the title compound the $\mathrm{N} 2-\mathrm{C} 5$ and $\mathrm{N} 4-\mathrm{C} 15$ distances are 1.424 (4) and 1.421 (4) $\AA$, respectively. These values are close to the $\mathrm{N}-\mathrm{C}_{\text {aryl }}$ bond length for p-nitrophenylmaleimide (Moreno-Fuquen et al., 2003) and are slightly smaller than the average value reported for nine N arylmaleimide derivatives (Miller et al., 2000). The benzene ring mean plane is rotated $56.2(1)$ and $52.9(1)^{\circ}$ with respect to the imide ring mean plane. These values are dictated probably by the weak hydrogen bond between an O atom of


Figure 1
The asymmetric unit of the title compound with the atomic labelling scheme. Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are shown as spheres of arbitrary radii.
the maleimide group and a C atom of the benzene ring. The rotation is smaller in the case of $p$-NPM, which has an angle of $42.98(5)^{\circ}$. This is consistent with the literature values, where other maleimides with bulky ortho substituents show angles of rotation greater than $80^{\circ}$. Other bond lengths and internal geometrical parameters of the title compound (Table 1) are similar to those in $p$-NPM. There are no significant intermolecular hydrogen bonds in the structure.

## Experimental

Reagents and solvents for the synthesis were obtained from Aldrich Chemical Co., and were used without additional purification. The title compound was prepared by taking equimolar quantities of $m$-nitroaniline and maleic anhydride in nitrobenzene and refluxing at 513 K for 3 h . The reaction product was filtered and washed with hexane and then it was dissolved in a mixture of ethyl acetate-hexane ( $15 \%$ hexane) in order to purify it by column chromatography. The solid was crystallized from chloroform, giving pale-yellow prisms with a melting point of 395 (1) K.

## Crystal data

| $\mathrm{C}_{10} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{4}$ | $Z=8$ |
| :--- | :--- |
| $M_{r}=218.17$ | $D_{x}=1.541 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Orthorhombic, $P n a 2_{1}$ | Mo $K \alpha$ radiation |
| $a=18.9815(6) \AA$ | $\mu=0.12 \mathrm{~mm}^{-1}$ |
| $b=6.6643(2) \AA$ | $T=123(2) \mathrm{K}$ |
| $c=14.8702(4) \AA$ | Prism, pale yellow |
| $V=1881.06(10) \AA^{3}$ | $0.40 \times 0.25 \times 0.07 \mathrm{~mm}$ |

## Data collection

| Enraf-Nonius CAD-4 | 1693 reflections with $I>2 \sigma(I)$ |
| :--- | :--- |
| $\quad$ diffractometer | $R_{\text {int }}=0.032$ |
| $\omega / 2 \theta$ scans | $\theta_{\max }=27.5^{\circ}$ |
| Absorption correction: none | 2 standard reflections |
| 4072 measured reflections | frequency: 150 min |
| 2238 independent reflections | intensity decay: $0.1 \%$ |

## Refinement

Refinement on $F^{2}$

$$
R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.045
$$

$$
w R\left(F^{2}\right)=0.119
$$

$$
S=1.05
$$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0754 P)^{2}\right. \\
& \quad+0.0433 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.39 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.27 \mathrm{e}^{-3}
\end{aligned}
$$

2238 reflections
289 parameters
H -atom parameters constrained

Table 1
Selected geometric parameters $\left(\AA{ }^{\circ}{ }^{\circ}\right)$.

| N2-C10 | $1.393(4)$ | C4-C5 | $1.394(5)$ |
| :--- | :--- | :--- | :--- |
| N2-C7 | $1.404(4)$ | C5-C6 | $1.392(4)$ |
| N2-C5 | $1.424(4)$ | C8-C9 | $1.318(6)$ |
| N4-C17 | $1.400(5)$ | C14-C15 | $1.383(4)$ |
| N4-C20 | $1.403(4)$ | C15-C16 | $1.386(5)$ |
| N4-C15 | $1.421(4)$ | $\mathrm{C} 18-\mathrm{C} 19$ | $1.329(6)$ |
|  |  |  |  |
| C10-N2-C5 | $124.8(3)$ | C20-N4-C15 | $124.8(3)$ |
| C7-N2-C5 | $124.2(3)$ | C2-C1-N1 | $118.8(3)$ |
| C17-N4-C15 | $124.8(3)$ | C16-C11-N3 | $118.4(3)$ |
|  |  |  |  |
| O1-N1-C1-C2 | $-6.9(5)$ | O5-N3-C11-C16 | $175.7(3)$ |
| C7-N2-C5-C6 | $116.7(4)$ | C20-N4-C15-C16 | $-47.8(5)$ |

In the absence of significant anomalous scattering, Friedel pairs were merged. H atoms were located in electron-density difference maps and subsequently treated as riding atoms, with $\mathrm{C}-\mathrm{H}=0.95 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: CAD-4 SDP (Frenz, 1978); program(s) used to solve structure: SHELXS86 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97.

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